

PFAS 101 Applied to the Solid Waste Industry



WDNR Solid Waste Interested Parties May 15, 2019

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Figure 3. Conceptual site model for landfills and WWTPs. (Source: Adapted from figure by L. Trozzolo, TRC, used with permission)

PFAS in the News







- Summary What's the Big Deal?
- PFAS What are they a little chemistry
- PFAS in Manufacturing
- Fate and Transport, and Toxicology,
- Regulatory Status
 - Emerging regs for long chain PFAS
 - Short chain PFAS regs, are they coming?
- PFAS in Landfills
- Treatment and Remediation

PFAS: What's the Big Deal



And Should We Be Concerned?

- **Regulations** Developing rapidly (e.g., 8 states added rules in 2018, 19 Total)
- **PFAS are Ubiquitous** Used in industry and in numerous commercial products since 1960s.
- **Public:** Public concern is growing (e.g., opposition groups using PFAS presence to argue against facilities)
- Standards in PPT: Varies by state. Water standards are typically 70 ppt, or less.
- **Highly Mobile:** Migration occurs easily and rapidly in groundwater, surface water and air.
- **Degradation:** No degradation of basic perfluorinated compounds
- Difficulties with PFAS:
 - Complicated: Sampling and analytical methods
 - Difficult to treat (\$\$\$) especially in leachate

What and Where are PFAS?



PFAS – Per- and Polyfluoroalkyl Substances. A large group of fluorinated compounds used for their stable, unreactive properties.

- Fire Fighting Foams:
 - Type B fire fighting foams Airports, refineries, etc.
- Textiles and Leather
 - Carpets, clothing, etc.
- Paper Products
 - Food contact materials
- Metal Plating Facilities
- Industrial Surfactants, Resins, Molds, Plastics
- Photolithography, Semiconductor Industry
- Biosolids
 - Paper mill sludge
 - Certain POTW sludge
- Landfills
 - Industrial wastes
 - Leaching from consumer products
 - Landfill fires, depending on whether foam was used.



A Little PFAS Chemistry

Quick Chemistry Lesson #1



- PFAS are Per and Polyfluoroalkyl substances
- <u>Per-fluoroalkyl substances</u>: can't be degraded, the fully fluorinated tail is very stable.



Quick Chemistry Lesson #2



- Examples of Poly-fluorinated compounds
- <u>Poly</u>-fluoroalkyl substances: non-fluorine atom (typically hydrogen or oxygen) attached to at least one carbon atom

8:2 Fluorotelomer Alcohol (8:2 FTOH)



Polyfluoroalkyl substances may also be degraded to perfluoralkyl substances (e.g. PFOS or PFOA, etc.)

The Evolution of PFAS As Shown in Fire Fighting Foams (AFFF)





Similar evolution in other industrial uses

PFAS in Landfills, Leachate, and Gas

PFAS in Landfill Leachate



- US Landfill Study (Lang et al., 2017) 95 samples from 18 landfills
 - 70 PFAS measured, 19 PFAS detected in >50% of samples
 - PFOS: 3 to 200 ppt
 - PFOA: 100 to 1,000 ppt
 - Total PFAS: 2,000 to 29,000 ppt

➢ 5:3 FTCA (precursor) dominant in most leachates: 400 to 15,000 ppt

- Canadian Landfill Study (Li, 2012) samples for 28 landfills
 - PFAS detections in all 28 samples
 - > PFOA detected in all samples, mean concentration of 439 ppt
- German Landfill Study (Busch, 2009) 22 German landfills
 - > 38 PFAS detected

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Total PFAS: 30.5 ppt to 13,000 ppt

National Estimate of Per- and Polyfluoroalkyl Substance (PFAS) Release to U.S. Municipal Landfill Leachate ES&T 2017

J. R. Lang, B. McKay Allred, J.A. Field, J.W. Levis, & Morton A. Barlaz



Michigan PFAS in Leachate & Effects on POTWs



Michigan Waste & Recycling Association/MDEQ (3/1/2019)

https://docs.wixstatic.com/ugd/6f7f77_9b845fefde8b4fd3b42e6a7bd321e21f.pdf

- 30 Mich. LFs, leachate analysis of PFOA & PFOS
 - PFOA range 3 to 800 ppt vs. MDEQ SW criterion 420 ppt
 - PFOS range 100 to 710 ppt vs. MDEQ SW criterion 11 ppt
- Landfills as PFOA/PFOS sources to WWTP are relatively minor
- Non-leachate sources contribute greater mass to WWTP influent than leachate.
- Study limitation: Analyzed for PFOA & PFOS only
 - No precursor analyses.
 - Precursors can be present in anaerobic conditions (i.e., in leachate), with degradation to PFOA and PFOS in aerobic treatment system.

Landfill Gas

- PFAS detected in:
 - Landfill Gas
 - Landfill gas condensate
 - Ambient air around landfill (and waste water treatment plants)
- Volatile Precursors
 - Some PFAS (e.g. fluorotelomer alcohols like 8:2-FTOH) have moderate volatility
 - These compounds can break down to form regulated PFAS in the environment
 - Significant PFAS (mostly FTOH) emissions (>1000 g/year) have been calculated from WWTPs and landfills (Ahrens et al, 2011)

Compound	Vapor Pressure (at 25-C)
2,3,7,8-TCDD	2E-07
Dieldrin	7E-04
PFOA	1.3
Naphthalene	10.6
8:2-FTOH	212
PCE	2550
TCE	9900
Benzene	12800



Fate and Transport and Toxicology

Fate and Transport of PFAS Landfills and WWTPs

Source Example





KEY 🙆 Atmospheric Deposition 💿 Diffusion/Dispersion/Advection 💿 Infiltration 💿 Transformation of precursors (abiotic/biotic)

Figure 3. Conceptual site model for landfills and WWTPs.

(Source: Adapted from figure by L. Trozzolo, TRC, used with permission)

Source: March 2018 ITRC Factsheet: Environmental Fate & Transport for PFAS, used with permission

PFAS Mobility







- Majority of US population exposed to PFAS
- Half-life = 2 10 years (humans)
- Prevalent in blood and urine samples baseline exists
- Can cross placental barrier exposure to developing fetus
- C8 Health project, 70,000 residents with drinking water exposure linked to serum-PFOA concentrations and variety of health outcomes







Reported Average Blood Serum PFOS Concentrations



- Most toxicology studies have focused on PFOA and PFOS
 - Conflicting study results in literature
- Has a low affinity to lipids and preferentially binds to proteins
 - Liver, kidney, muscle and blood
- Biological Fate in Humans



- Dec 2018 USEPA IRIS announced 5 PFAS (PFNA, PFBA, PFHxA, PFHxS, PFDA) will be reviewed for toxicity assessment
 - Some of these are the short chain PFAS compounds that form the bases for Modern AFFF



Regulatory Status What is Occurring? – What to Expect?



U.S. EPA's Recent Actions



EPA PFAS Action Plan: An Overview

- PFOA & PFOS proposed to be listed as hazardous substances under CERCLA
- Development of MCLs for PFOA & PFOS
- Expand monitoring of PFAS in UCMR5
- Consider PFAS for Toxics Release Inventory
- Develop risk communication toolbox
- Develop toxicity standards for select PFAS
- Develop new analytical methods for PFAS
- Evaluate treatment options for PFAS
- PFOA & PFOS groundwater cleanup recommendations



EPA wants to close gaps on PFAS science ASAP.

EPA Announces Most Comprehensive Cross-Agency Action Plan in History of EPA for PFAS

- April 25, 2019 Draft Groundwater Remediation Standards for public comment
 - 40 ng/L for a screening level
 - 70 ng/L for the preliminary remediation goal (PRG)



Comprehensive Environmental Response, Compensation, and Liability Act Provides broad authority to federal government (including natural resource trustees) to respond directly to releases or threatened releases of **hazardous substances** that may endanger public health or the environment.

- At present, CERCLA lists ~ 800 hazardous substances
- EPA has initiated the regulatory development process for listing PFOA & PFOS as CERCLA hazardous substances. This would allow:
 - EPA to require responsible parties to carry out and/or pay for response actions
 - ✓ Private parties to seek cost recovery for their response actions

Potential for CERCLA reopeners at 5-year review

Potential impacts to existing state cleanup programs for PFAS

Regulatory Status – The States





- Enforceable limits
- Non-enforceable advisories, notification levels, etc.

No standards

PFAS: The Rapidly Changing Regulatory Landscape

Follow ITRC for updates

https://pfas1.itrcweb.org/fact-sheets/

	Year First		Promulgated				Other
State	Listed	Туре	?	PFOA	PFOS	PFNA	PFAS
Alacka	2016	GW	Υ	0.400	0.400		Ν
	2018	DW GW/SW	N	0.070	0.070	0.070	Y
California	2018	DW	Ν	0.014	0.013		Ν
Colorado	2018	GW	Y	0.070	0.070		Ν
Connecticut	2016	DW/GW	Ν	0.070	0.070	0.070	Y
Delaware	2016	GW	N	0.070	0.070		Ν
Delaware	2016	GW	Ν	0.070	0.070		Y
lowa	2016	Protected GW	Y	0.070	0.070		Ν
	2016	Non-protected GW	Y		1		
Maine	2016	DW	Ν	0.070	0.070		Ν
	2018	GW	Ν	0.400	0.400		Y
	2016	GW	Ν	0.120	0.120		Y
	2016	SW/RW	Ν	0.170	0.300		Y
Massachusetts	2018	DW	0	0.070	0.070	0.070	Y
Michigan	2015	SW	Y	0.420	0.011		Ν
	2018	DW/GW	Y	0.070	0.070		Ν
Minnesota	2017	DW/GW	O/N	0.035	0.027		Y
	2017	DW/GW	O/N	0.035	0.027		Y
	2017	DW/GW	O/N	0.035	0.027		Y
Nevada	2015	DW	N	0.667	0.667		Y

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PFAS: The Rapidly Changing Regulatory Landscape



	Year First		Promul	-			Other
State	Listed	Туре	gated ?	PFOA	PFOS	PFNA	PFAS
New Hampshire	2016	GW	Y	0.070	0.070		N
New Jersey	2018	GW	Y			0.013	Ν
	2018	DW	Y			0.013	Ν
	2017	DW	0	0.014			Ν
	2018	DW	0		0.013		Ν
North Carolina	2006	GW	Y	2			Ν
	2017	DW	Ν				Y
Oregon	2011	SW	Y	24	300	1	Y
Pennsylvania	2016	GW	Ν	0.070	0.070		Ν
Rhode Island	2017	DW/GW	Y	0.070	0.070		Ν
Texas	2016	GW	Y	0.290	0.560	0.290	Y
Vermont	2018	DW/GW	Y	0.020	0.020	0.020	Y
	2016	GW	Y	0.010	0.010	0.010	Y



Sampling and Evaluation



PFAS Sampling Dos and Don'ts A few examples from TRC's SOP



WHAT SHOULD I AVOID?	USE INSTEAD			
Equipment with Teflon ® (e.g., bailers, tubing, parts in pump) during sample handling or mobilization/demobilization	 High density polyethylene (HDPE) or silicone tubing/materials in lieu of Teflon® 			
Low-density polyethylene (LDPE) or glass sample containers or containers with Teflon- lined lids A few examples only. A complete	 HDPE or polypropylene containers for sample storage listHDPE or polypropylene caps 			
PFAS Sampling. Tyvek® suits and waterproof boots	 Clothing made of cotton preferred Boots made with polyurethane and polyvinyl chloride (PVC) 			
Waterproof labels for sample bottles	 Paper labels with clear tape 			
Sunscreens, insect repellants	 Products that are 100% natural, DEET 			
Sharpies	 ✓ Ballpoint pens 			
Aluminum foil	✓ Thin HDPE sheeting			

Separating Impacts



Example: Fingerprinting Multiple Sources of Fire Fighting Foam



Blue – PFOS Grey PFHxA

Green- PFHxS

Orange PFOA

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Treatment and Remediation



PFAS Waste Water Treatment

PFAS Remediation Challenges

- Low Volatility (rules out stripping)
- Moderate solubility
- Strength of C-F Bond
- Treatment efficiency must be very high because of low (ppt) remediation objectives

Water Treatment Technologies

- Sorption/Ion Exchange
 - Carbon (can be effective for some PFAS, but can be inefficient)
 - Ion Exchange Resins (costly)
 - Need to remove all other organics before PFAS treatment





PFAS Remediation Alternatives

Ex-Situ Technologies

- Sorption/Ion Exchange
 - Carbon (can be effective for some PFAS, but can be inefficient)
 - Ion Exchange Resins (costly)
- Emerging technologies:
 - Reverse Osmosis
 - Membrane filtration
 - Thermal Treatment
 - SAFF Surface Activation Foam
 Fractionation



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In Situ Technologies

- Emerging(?) technologies:
 - Carbon injection
 - PRB or Source Area
 - Electro-Chemical
 Oxidation
 - ART In-Well Circulation
 System



Thank you

Questions?

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